

# Rhodium(III)-Catalyzed Selective Monoarylation of $\beta$ or $\gamma$ C(sp<sup>3</sup>)-H Bonds Assisted by a Trimethylpyrazole Group

Chunchen Yuan, Guangliang Tu, and Yingsheng Zhao\*

Key Laboratory of Organic Synthesis of Jiangsu Province, College of Chemistry, Chemical Engineering and Materials Science, Soochow University, Suzhou 215123, China

Supporting Information

**ABSTRACT:** The selective arylation of unactivated  $\beta$  or challenging  $\gamma$  primary and secondary  $\beta$ -C(sp<sup>3</sup>)-H bonds has been developed with a Cp\*Rh(III) catalyst assisted by a trimethylpyrazole group. A rarely reported six-membered rhodacycle has been identified in rhodium-catalyzed C(sp3)-H activation reactions. Preliminary mechanistic studies have revealed that a concerted metalation-deprotonation pathway might be involved in the C-H activation step.

ver the past decades, transition-metal-catalyzed C-H activation reactions have seen rapid development and now provide straightforward approaches to many important synthetic units. 1-5 Among the reported procedures, rhodium-(III)-catalyzed C-H functionalization has attracted considerable attention in recent years due to its high selectivity, good functional group tolerance, and high efficiency.6 To date, numerous of Rh-catalyzed C(sp<sup>2</sup>)-H functionalizations have been developed, leading to carbon-carbon and carbonheteroatom bond formation. Despite these great achievements, only limited reports have been concerned with the rhodium-(III)-catalyzed C(sp<sup>3</sup>)-H functionalization, which may be due to an incomplete understanding of pertinent mechanistic aspects.

In 2010, Glorius' group reported pioneering work on the rhodium(III)-catalyzed allylic C(sp<sup>3</sup>)-H activation of enamines in coupling with alkynes to form pyrazoles.<sup>8</sup> Later, Wang's group showed 8-methylquinolines to be suitable substrates for rhodium(III) catalysts in activating C(sp<sup>3</sup>)-H for alkenylation reactions. Since then, several examples of selective functionalization of C(sp<sup>3</sup>)-H bonds with Cp\*Rh(III) as catalyst have been developed by the groups of Glorius, 10 You, 11 and Li. 12 However, the substrates have been limited to 8-methylquionline-type benzylic C-H bonds or oxime-directed  $\beta$ -Me C-H bonds. There have been few examples of functionalization of unactivated primary and secondary C(sp<sup>3</sup>)-H bonds. In particular, to the best of our knowledge, there has been no example of rhodium-catalyzed  $\gamma$ -C(sp<sup>3</sup>)–H bond functionalization, presumably because an unstable six-membered rhodacycle would need to be formed during the catalytic cycle. Herein, we report a rhodium(III)-catalyzed selective monoarylation of C(sp<sup>3</sup>)-H bonds assisted by a trimethylpyrazole group. Preliminary results have revealed that rhodium(III) can not only promote functionalization of  $\beta$ -Me C-H bonds but also of less activated methylene C(sp<sup>3</sup>)-H bonds. Further study has revealed that primary  $\gamma$ -C(sp<sup>3</sup>)-H bonds are compatible,

whereby a rarely reported six-membered Rh-C(alkyl) intermediate may be invoked for the first time.

Although pyrazole-directed  $C(sp^2)$ -H functionalizations have been well explored and various transition metals such as Rh, 13 Pd, 14 and Ru 15 have been demonstrated to be effective catalysts, there have been few examples of pyrazole-directed C(sp<sup>3</sup>)-H activations. It is likely that the weak coordinating ability of pyrazole hinders C-H transformations. 16 Recently, Yu's group 17 has overcome this problem by employing monoprotected amino acids as ligands, which are well-known to promote C-H activation reactions with palladium catalysts. On the basis of our experience in developing new directing groups, 18 the electron density is known to greatly influence the efficacy of transition-metal-catalyzed C-H activations. We speculate that the pyrazole skeleton may provide a perfect coordinating center for C(sp<sup>3</sup>)-H activation if its structure is slightly modified. Thus, 1-alkylpyrazole derivatives might be suitable substrates for rhodium-catalyzed C(sp³)-H activation to form important synthetic units. Further investigation indicated that these 1-alkylpyrazoles could be easily prepared from an alkyl bromide, <sup>19</sup> an alcohol<sup>20</sup> and a (trifluoromethyl)sulfonyloxy-protected alcohol.<sup>21</sup>

With these conditions in mind, we initially treated 1isopropyl-1H-pyrazole (1a) with a triarylboroxine (2) in the presence of [Cp\*RhCl<sub>2</sub>]<sub>2</sub> (5 mol %), Ag<sub>2</sub>CO<sub>3</sub> (2 equiv), and AgSbF<sub>6</sub> (0.2 equiv) in toluene for 24 h (Scheme 1, 3a). Unfortunately, the arylated product 3a was only observed by LC-MS in <3% yield. Next, pyrazoles bearing electronwithdrawing functional groups were tested (3a). However, these did not give the desired products, and only the starting material was recovered. It is probable that weakly coordinating pyrazoles have no assisting ability in promoting C-H activation. Monomethyl-substituted pyrazoles were further

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# Scheme 1. Scope of Rhodium-Promoted $C(sp^3)$ -H Activation<sup>a</sup>

<sup>a</sup>FG: functional group.

screened. As we expected, monoarylated products were obtained in up to 18% yields (3b,c). Encouraged by these results, 1-isopropyl-3,5-dimethylpyrazole (1d) was tested. To our great delight, the arylated product 3d was obtained in 38% yield. The 1-isopropyl-3,4,5-trimethylpyrazole was further explored, affording the monoarylated product 3e in a rather good yield of 56%. When substrate 1h was scanned, less than 10% of the arylated product (3f) was obtained.

With these preliminary results in hand, we next explored various oxidants, solvents, and additives to accomplish this  $\beta$ - $C(sp^3)$  arylation reaction (see the Table S1). The results revealed that the additives NaOAc, PivOH, (n-BuO)<sub>2</sub>PO<sub>2</sub>H, and 1-Ad-OH all had a promoting effect. A satisfactory isolated yield of 3g of 85% was achieved when PivOH was used as the additive. In regard to the role of these additives in palladiumcatalyzed C(sp<sup>3</sup>)-H activation, a concerted metalationdeprotonation (CMD) pathway might be involved in activating the  $C(sp^3)$ -H bonds. <sup>22</sup> Several oxidants, such as  $Cu(OAc)_2$ , BQ, K<sub>2</sub>S<sub>2</sub>O<sub>8</sub>, and O<sub>2</sub>, were screened in place of Ag<sub>2</sub>CO<sub>3</sub>. However, none of them gave the desired products in good yields. It is worth mentioning that the monoarylated product 3g was the only arylated product. We surmise that the rhodium-(III) complex interacted weakly with the arenes, which precluded secondary C-H activation, while the steric effect is also a possible reason for the absence of a secondary arylation. In addition, palladium acetate was also tested in combination with aryl iodides as coupling partners. However, mono- and diarylated products were obtained unselectively, highlighting the unique properties of the rhodium(III) catalyst system in the C-H activation.

A wide range of triarylboroxines were treated with 1-isopropyl-3,4,5-trimethylpyrazole (1-isopropyl-TMP) (1g) to explore the functional group tolerance of this rhodium-catalyzed C(sp³)—H arylation (Scheme 2). Arylboron reagents (2) bearing *para* or *meta* substituents gave the corresponding products in moderate to good yields (53–85%). A wide variety of functional groups, such as MeO, F, Cl, Br, CF₃, and CO₂Me, were well tolerated in this transformation (3g–p). Multiply substituted arylboron reagents also reacted well, affording the corresponding monoarylated products in good yields (3q,r,t). Importantly, condensed-ring boron reagents also performed

Scheme 2. Scope of Arylboron Reagents in the Rhodium-Catalyzed C(sp<sup>3</sup>)-H Arylation\*

\*Reactions were performed with 1 (0.2 mmol) and 2 (2 equiv) in 1 mL of toluene.  $^a[Cp*RhCl_2]_2$  (10 mol %).  $^b150$  °C

well, providing the corresponding products in good yields (3s,u,v).

We next investigated various 1-alkyl-TMPs under the optimal conditions (Scheme 3). The substrates 4a and 4b were both selectively arylated at the  $\beta$ -Me position, providing the monoarylated products (5a,b) in good yields. 1-Ethyl-TMP also reacted smoothly, leading to monoarylated product 5c in 78% yield. Moreover, 1-cyclobutyl-TMP, 1-cyclopentyl-TMP, and 1-cyclohexyl-TMP all gave the monoarylated products in moderate to good yields, although methylene units are generally less reactive in C-H activation. However, the substrates of n-propyltrimethylpyrazole (4g) could not be tolerated under standard reaction conditions, and only starting material was recovered. The product of 5c, which contained  $\beta$ secondary benzylic C-H bonds, was treated with triarylboroxine 2a, but it failed to give any  $\beta$ -arylated product. The substrate 4h was unreactive in the reaction, affording the arylated products in less than 10% yield.

In all previously reported rhodium(III)-catalyzed  $C(sp^3)$ –H activations, a five-membered Rh–C(alkyl)-cyclic intermediate has been invoked. To the best of our knowledge, a six-membered Rh–C(alkyl)-cyclic intermediate has not hitherto been observed in a Cp\*Rh(III)-catalyzed  $C(sp^3)$ –H activation reaction. Inspired by palladium-catalyzed C–H activation reactions, in which a six-membered Pd–C(alkyl)-cycle is readily formed when the C–H activation step proceeds by a CMD pathway, we speculated that the modified pyrazole-directed rhodium-catalyzed  $C(sp^3)$ –H activation may also facilitate  $\gamma$ - $C(sp^3)$ –H arylation. Thus, 2,2-dimethylpropane-TMP was examined under the standard conditions. As expected, the monoarylation reaction proceeded well with different triarylboroxine reagents (6a,b). For example, both

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# Scheme 3. Scope of Arylboron Reagents in Rhodium-Catalyzed C(sp<sup>3</sup>)-H Arylation Reactions\*

\*Reactions were performed with 4 and  $(ArBO)_3$  (2 equiv) in 1 mL of toluene.  ${}^a[Cp*RhCl_2]_2$  (10 mol %), 150 °C.

electron-rich and electron-deficient substituted triarylboroxine reagents were tolerated in the transformation, indicating the potentially wide scope of these reagents. Isopentyl-TMP gave monoarylted **6c** in 58% yield.

Gram-scale reactions were easily performed with a range of triarylboroxine reagents under the standard reaction conditions (Scheme 4a). More importantly, we demonstrated that the trimethylpyrazole coordinating center could be easily removed from product 3j under basic conditions at room temperature, affording the important product anethole, which is widely used as a flavoring agent (Scheme 4b). This result implied that not

# Scheme 4. Further Study on This Rhodium-Catalyzed C(sp³)—H Arylation Reaction

only can various pyrazole derivatives be prepared from simple alkyl halides but also a facile approach to alkenes from alkyl halides via a transitive directing center has also been developed.

Because mono- and diarylated products were inevitably obtained when we treated 1g with aryl iodides in the presence of a palladium catalyst, the product 5c was further treated with the aryl iodide using the palladium catalyst to form the heterodiarylated product. To our delight, the arylation occurred selectively at the  $\beta$ -C(sp<sup>3</sup>)-H position rather than at the  $\delta$ -C(sp<sup>2</sup>)-H position, indicating that a five-membered Pd-C(alkyl) intermediate is more stable than a seven-membered Pd-C(arene) intermediate. We also treated 5c with triarylboroxine 2a under the optimal reaction conditions, but it proved to be completely unreactive, which explains why the diarylate could not be observed in the Cp\*Rh-catalyzed C(sp³)-H arylation and is a clear reason why a diarylated product was not obtained under the Rh(III)-catalyzed conditions. We proposed that the rhodium(III) complex might have interacted weakly with the arenes, which prevented secondary C-H activation. In addition, steric effects might also be a reason for the absence of a secondary arylation.

Several experiments were performed to gain more insight into the mechanism. Hydrogen—deuterium-exchange experiments were explored with the rhodium catalyst or palladium acetate. A deuterated product was never observed, indicating that C(sp³)—H bond activation is irreversible (see the SI). Intermolecular kinetic isotope effect (KIE) experiments were carried out (Scheme 5a), which gave a consistent KIE of 2.3.

### Scheme 5. Preliminary Mechanism Study

a) KIE study 
$$FG \\ Me \\ N \\ N \\ N \\ N \\ + 2a \\ \hline \begin{array}{c} \{\{Cp^*RhCl_2\}\}_2 (5 \text{ mol } \%) \\ Ag_2CO_3 (2 \text{ equiv}) \\ Piv-OH (10 \text{ mol } \%) \\ Ag_3EF_6 (20 \text{$$

Control experiments revealed that the additives AgSbF<sub>6</sub> and PivOH both had a promoting effect (Scheme 5b). To understand the role of these additives, a catalyst precursor was prepared by the reaction of [Cp\*RhCl<sub>2</sub>]<sub>2</sub>, AgSbF<sub>6</sub>, and NaOAc in acetonitrile, affording a complex in the form of deepred prisms. Its structure was confirmed by single-crystal X-ray diffraction analysis (Scheme 5c). Complex A was further utilized as the catalyst instead of [Cp\*RhCl<sub>2</sub>]<sub>2</sub>, AgSbF<sub>6</sub>, and PivOH, and the desired product 3g was obtained in 61% yield

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(Scheme 5d). This result implied that complex A is an active species in the catalytic cycle.

In conclusion, we have developed a rhodium-catalyzed site-selective monoarylation of both methyl and methylene at either the  $\beta$  or  $\gamma$  position by employing a modified pyrazole as a transitive coordinating center. Preliminary mechanistic studies have revealed that a CMD pathway might be involved in the C-H activation step. This may pave the way for a comprehensive understanding of the mechanism of rhodium-(III)-promoted  $C(sp^3)$ -H activation.

### ASSOCIATED CONTENT

## **S** Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b03522.

Experimental procedures and <sup>1</sup>H and <sup>13</sup>C NMR spectra for new compounds (PDF)

X-ray data for rhodium catalyst precursor (CIF)

#### AUTHOR INFORMATION

#### **Corresponding Author**

\*E-mail: yszhao@suda.edu.cn.

ORCID ®

Yingsheng Zhao: 0000-0002-7977-5284

Notes

The authors declare no competing financial interest.

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